

Nutrient recovery from wastewater streams: new perspectives and technical solutions

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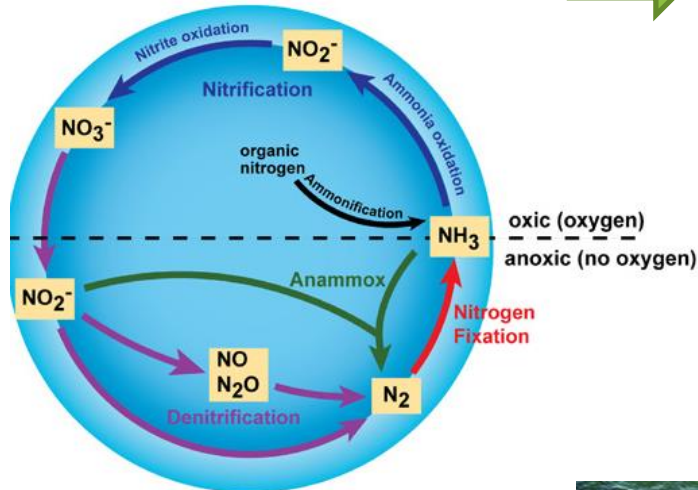
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Leverage from
the EU
2014–2020

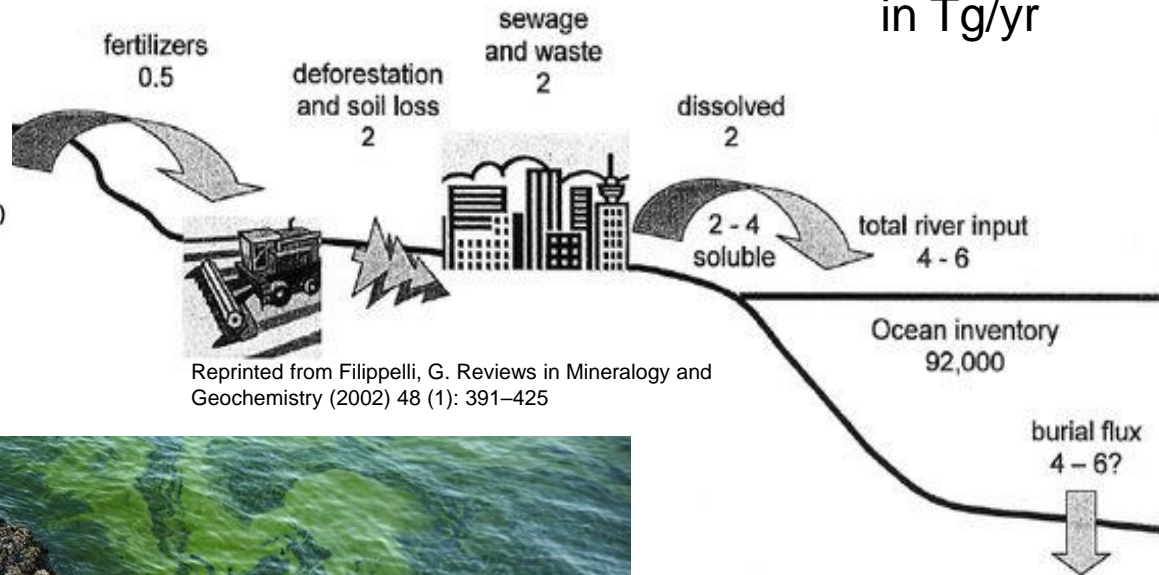


Motivation

The human alteration of nitrogen and phosphorus global natural cycles  heavily unbalanced



Reprinted from Bernhard, A. (2010) The Nitrogen Cycle: Processes, Players, and Human Impact. Nature 3(10):25



Price?
annually

- U.S. **\$2.2 billion** (Dodds et al. 2009)
- UK **\$160 million** (Pretty et al. 2003)
- Baltic sea **€ 200 million** (HELCOM)



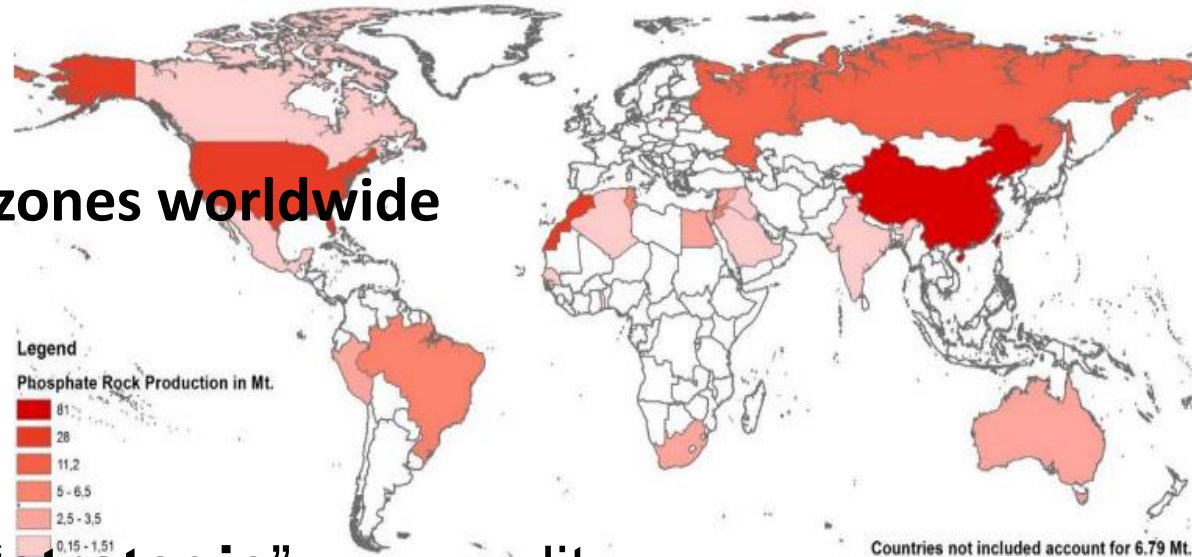
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Motivation

- >400 coastal **dead zones worldwide**



- Phosphorus is a **“strategic”** commodity located in just few countries and deposits included into the list of critical materials in **2014** up 80% of mined phosphorus is lost the extraction process

- Recovery could be valuable in near future will be depleted within the next 45–100 years as discharge limits become stricter

- Possibility to cover **50% of a P-market**

Motivation

- developing less energy-demanding process for nitrogen capturing – Haber-Bosch process accounts for approximately **1% of global energy consumption**;
- to **reduce GHG emission**: substantial concomitant as Haber-Bosch process and WWTP facilities (biological treatment, 3% of all GHG globally);
- to reduce nutrient leaching into natural water bodies
- transition to **circular economy**

Benefits

Renewable fertilizers or industry feedstocks

Protecting and improving water quality

Improving operation and performance at WWTP

Improving food security and social equity

Business opportunities

Nutrient-contaminated streams



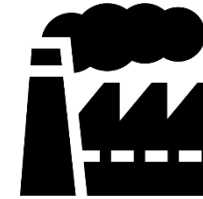
Agriculture, fishery, forestry

- Facultative lagoons
- Air purification cattle farms
- Overfertilized fields
- Aquaculture operations
- Ditches and peat bogs



Municipal and urban wastewaters

- Run-off and storm waters
- Wastewater treatment plants



Industrial waste- and processing waters

- Reject waters of biogas stations
- Landfill leachates
- Textile industry processing waters
- Paper mills grey waters
- Food industry (meat and beverage)

Low-laden or High-laden with nutrients

Diffuse or centralized source

Complexity of matrix

Emerging pollutants

Energy demand

Public perception and acceptance

Legislation

N/P ratio

Nutrients in municipal wastewater

Globally

380 billion m³/a of wastewater

Major nutrients in wastewater, worldwide, annually:

16.6 million metric tonnes of N

6.3 million metric tonnes of K

3 million metric tonnes of P

Recovered these nutrients could offset of global demand:

14.4% for N;

6.8% for P

18.6% for K

theoretically generated **revenue globally:**

\$9.0 billion from the recovery of N,

\$2.3 billion from P,

\$2.3 billion from P.



Energy, chemical consumption, sludge production
critical environmental benefits such as minimising eutrophication

Nationally*

0.5 billion m³/a of wastewater

Nitrogen load in sewage systems **32 670** ton/a and **66%** of it is removed and **lost** in current treatment processes, **rest** in inland waters

Phosphorous load in sewage systems **4 300** ton/a and **3%**** of this was recycled

In 2017, a total of app.155 tonnes of **P** and app. 11,090 tonnes of **N** were discharged from municipal wastewater treatment plants.

POSSIBLE BUSINESS CASES

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European Union
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SOLUTION is

Combination of mature techniques

Adsorption ➤

**New geopolymer adsorbent from low-cost sources:
unique properties and steady quality**

+
Desorption ➤

adsorption-desorption cycles without reloading

=
Concentration ➤

the max enrichment factor with min reagent use

**Air stripping
OR
Membrane separation**



**Independence from initial concentration
pH already suits for this technologies
Twice lower temperature could be used
Minimize the dimensions of the setup
Lower energy consumption**

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Design of the Treatment Process

Objects

- **Run-off waters and effluents**
Diluted solutions with complex matrix < 0.2 g N/L

1 stage

- **Nitrogen removal**
• Adsorption by new geopolymers

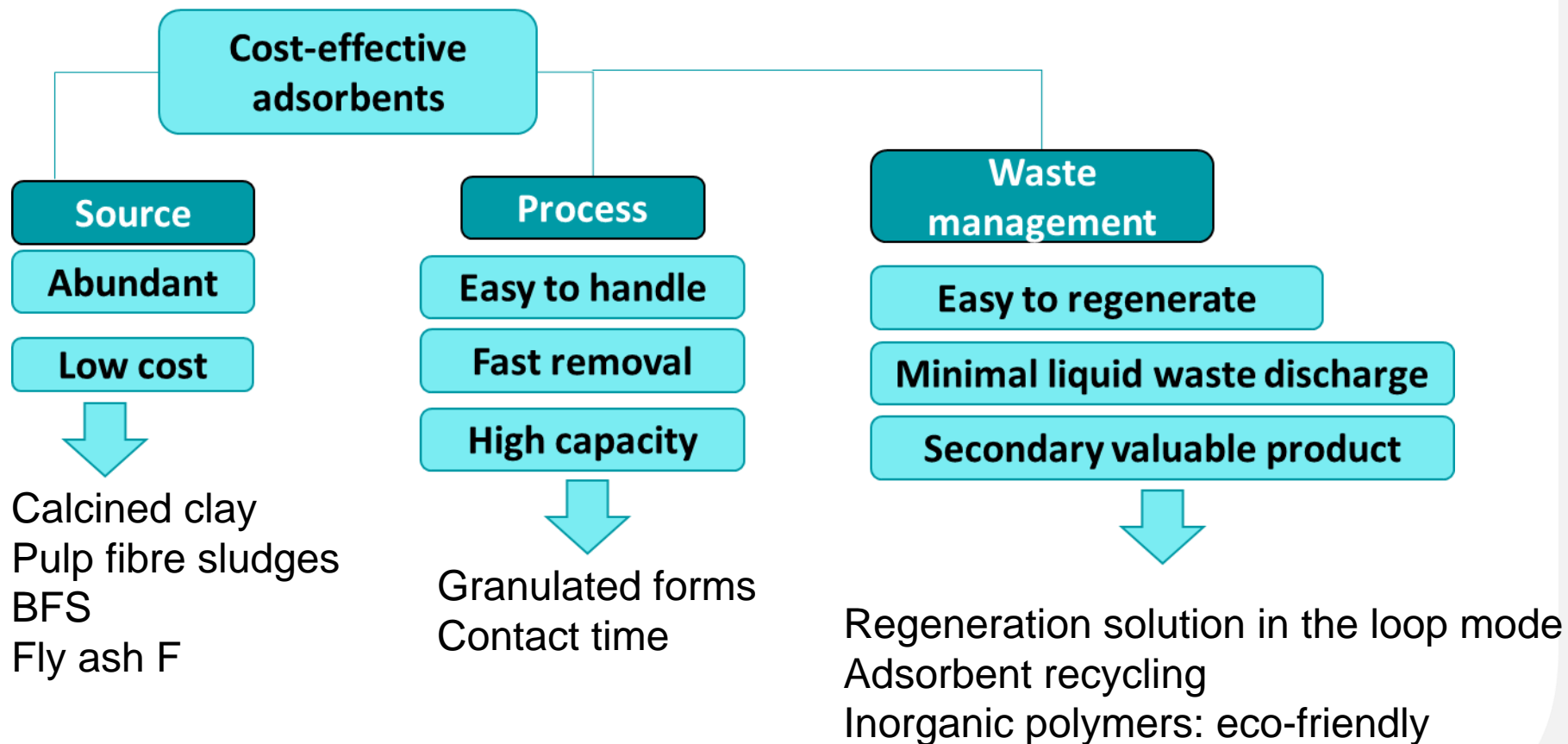
2 stage

- **Nitrogen preconcentration**
• Desorption with a regeneration solution

3 stage

- **Nitrogen recovery**
• Air-stripping or membrane technology for the regeneration solution purification

Stage 1: Design of cost-effective adsorbents



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Geopolymers for ammonium concentration

Raw material for geopolymer preparation:

- paper mill sludge contained kaolinite as paper additive; **FS MKGP**
- calcined kaolinite (Luukkonen et al. 2016)

To recover clay and transform it to form needed, paper mill sludge was ignited, and the remaining kaolinite calcined at 750°C.

- commercially available metakaolin (Aquaminerals Finland Oy); **MKGP**

To prepare reference geopolymer material, metakaolin was geopolymerized in accordance with standard procedure (Luukkonen et al. 2016).

Adsorbent	Capacity mg/g	Regeneration
MKGP	1,3	5M NaCl, pH 12
FS MKGP	3,7	5M NaCl, pH 12 + acetic acid
Zeolite (Spain)	2,2	5M NaCl, pH 12 + acetic acid
Zeolite (China)	1,3	5M NaCl, pH 12
Zeolite (Bulgaria)	3,7	5M NaCl, pH 12 + acetic acid

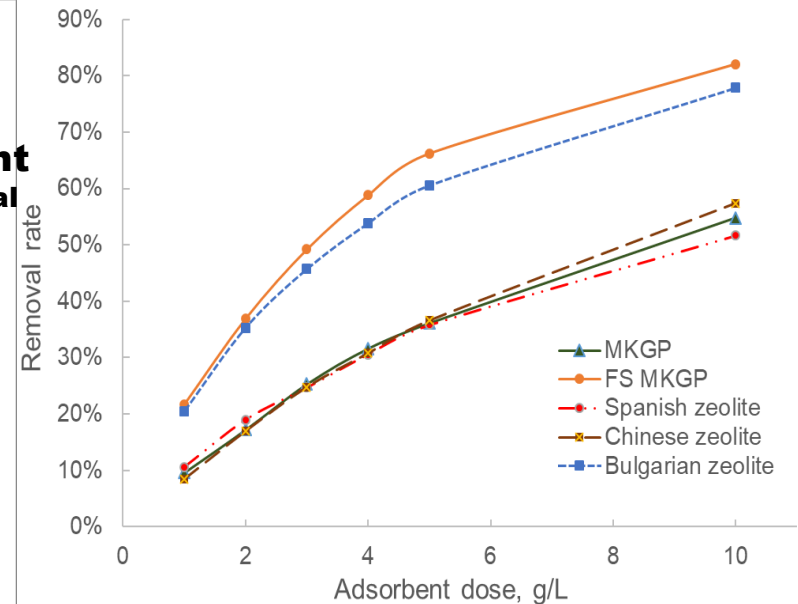
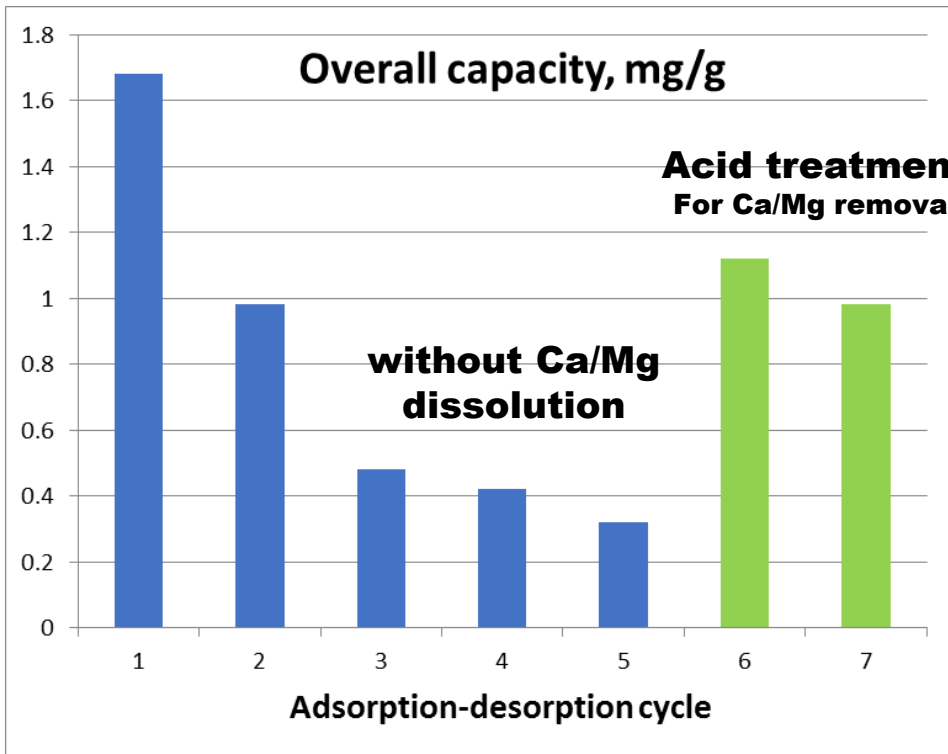
Regeneration study

Sample	Run-off	Lake water	Mine water		WWTP effluents	
			Point 1	Point 2	Before	After
Adsorbent						
MKGP	68 %	64 %	44 %	45 %	54 %	52 %
FS MKGP	92 %	95 %	69 %	72 %	82 %	81 %
Spanish zeolite	61 %	58 %	32 %	38 %	51 %	56 %
Chinese zeolite	55 %	58 %	17 %	21 %	57 %	57 %
Bulgarian zeolite	87 %	84 %	52 %	60 %	77 %	78 %

adsorbent dose
10 g/L

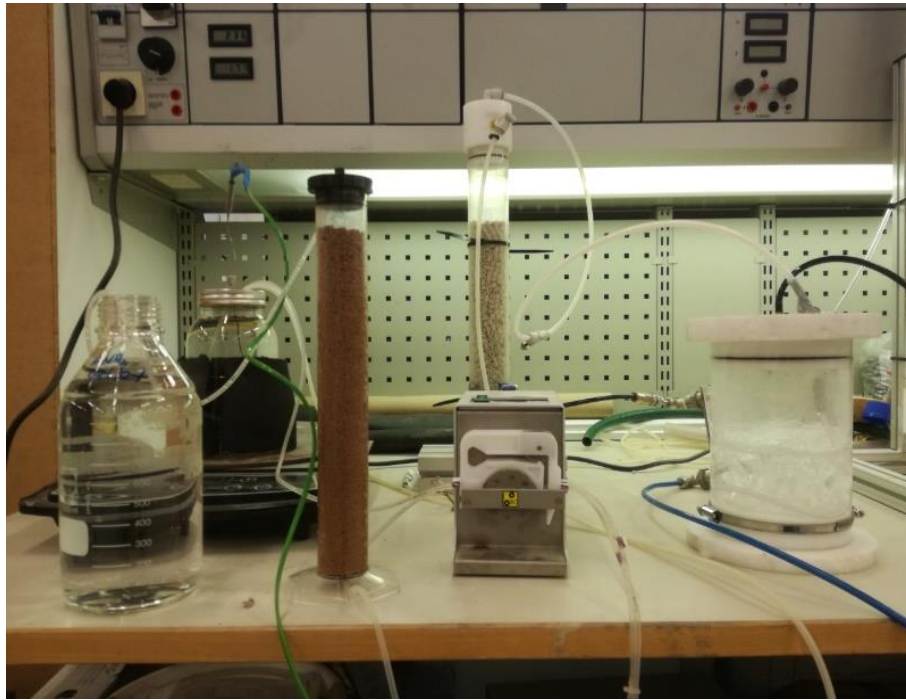
grain size
63-125 µm

contact time 3h



Lab-scale demonstrations and pilotings

Geopolymer adsorbents:
Adsorption-desorption cycles were piloted on local WWTP Kajaanin Vesi



Lab-scale setup for air-stripping evaluation:
adsorption and stripping cycles analysis
alkali and acid dosage
optimal temperature testing
liquid and air supply rates
hydraulic loading optimization

Results

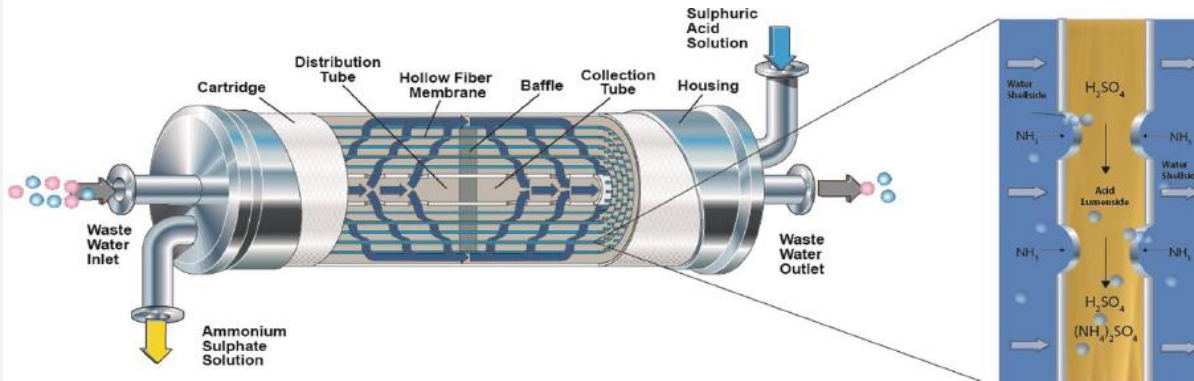
Temperature $45 \pm 5^\circ\text{C}$ was enough to reach conversion rate 91 %.

After the regeneration solution was purified, it was used over 5 times for desorption procedure.



Lab-scale demonstrations and pilotings

Transmembrane Chemical Absorption



Results

3M Liqui-Cel[®] membrane contactor provided by manufacturer for piloting on WWTP

Receiving acids: phosphoric or sulfuric



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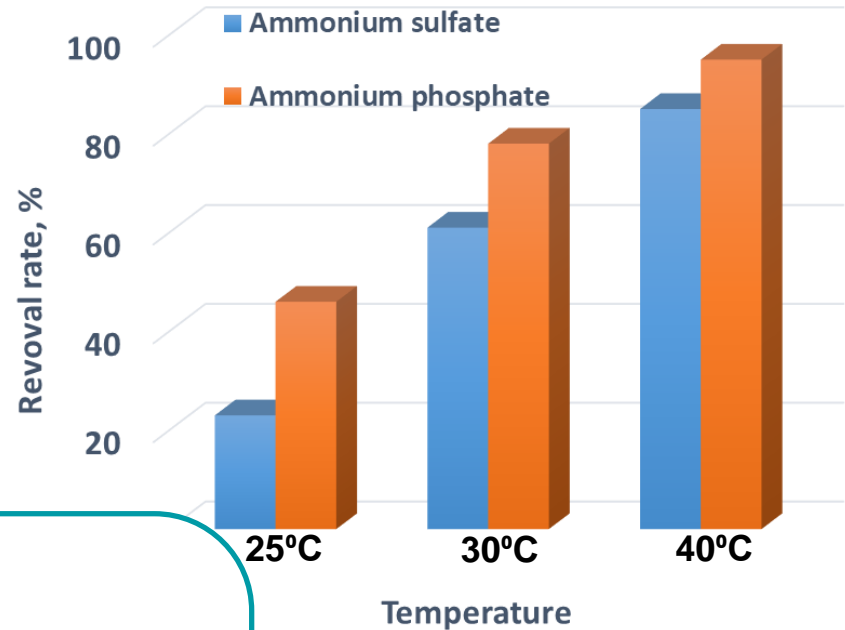


Recovery of ammoniacal nitrogen Transmembrane Chemical Absorption

Final product:
ammonium sulfate

ammonium
mono(di)phosphate

- **Optimized conditions:**
 - 100 L/h shellside flow with $\text{pH} \geq 11$
 - 60L/h lumenside feed flow with $\text{pH} \leq 5,6$
 - working temperature - 40°C.
- **Lumenside absorption liquid:**
 - Sulfuric or phosphoric acids (tech.) up to 5%.



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Lab-scale demonstrations and pilotings



First piloting for
WaterPro
at Kajaani Vesi

- Prebooked for Karelia CBC piloting (SUSWAM, REMAC)

Nutrient removal from run-off waters

What else could be beneficial?

Unique properties



Buoyancy

High porosity

Lightness but hardness

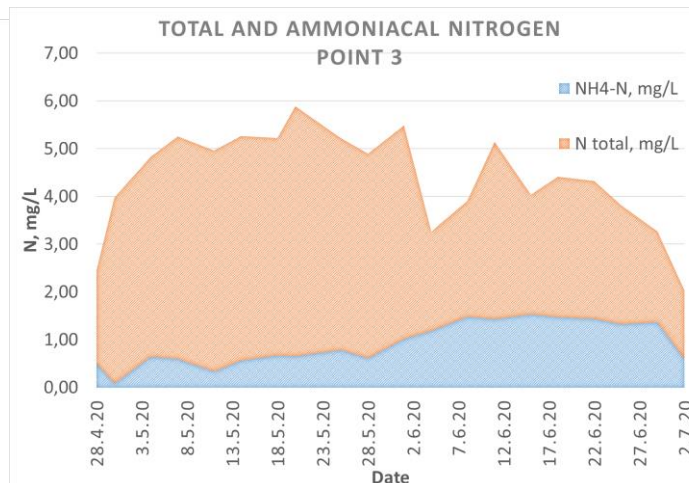
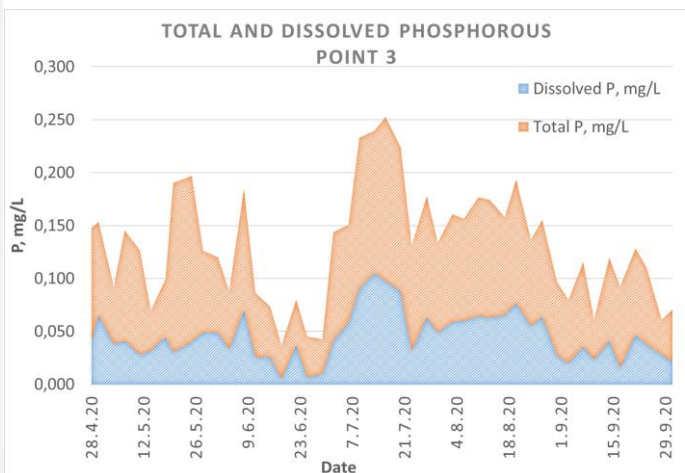


Work co-funded by of Maa- ja vesitekniiikan tuki (№ 13-8271-17).

Lab-scale demonstrations and pilotings



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Lab-scale demonstrations and pilotings

**After 10 weeks
of piloting**

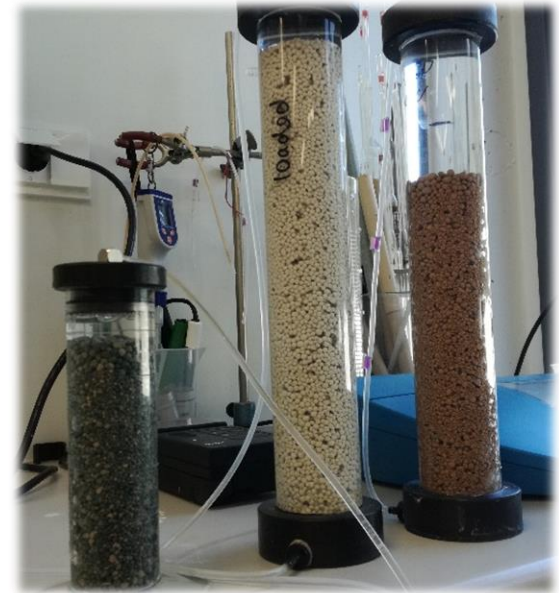
Total P less than 0,03 mg P/L
PO₄-P less than 0,01 mg P/L

↓ ~60%

Total N less than 1 mg N/L
NH₄-N less than 0,05 mg N/L

↓ ~80%

				Adsorbent		
				MKGP per 40 kg	LECA-MKGP per 10 kg	FSMKGP
Chemicals	per 1 ton	per kg				
BFS		55	0,055	-		
NaSi		100	0,100	0,5	0,84	0,5
NaOH		180	0,180	1,08	0,45	1,08
MK		200	0,200	8	1,5	
LECA, m3		60	0,060		0,6	
Perlite, m3		35	0,035			
Cork		1200	1,200			
kaolinite clay		50	0,050			
Supplies	1h					
water		2,25	0,003			
electricity		15	0,177	2,66	2,66	14,66
				12,24	6,05	16,24
		EUR per 1ton		306	151	406



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Thank You for
your
attention!

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